

Asymmetric Addition Reaction of Organozinc Reagents to Nitrones Using a Catalytic Amount of External Chiral Auxiliary

Yutaka Ukaji,* Yuuichi Kenmoku, and Katsuhiko Inomata*

Department of Chemistry, Faculty of Science, Kanazawa University, Kakuma, Kanazawa, Ishikawa 920-11, Japan

Abstract: Catalytic asymmetric addition reactions of dialkylzinc to nitrones are realized; *i.e.*, in the presence of a catalytic amount of bromomagnesium (2*S*,3*R*)-4-dimethylamino-1,2-diphenyl-3-methyl-2-butoxide, dialkylzincs were reacted with 3,4-dihydroisoquinoline *N*-oxide derivatives to give the corresponding 1-alkylated hydroxylamines enantioselectively. In order to achieve higher stereoselection, addition of bromomagnesium triphenylmethoxide was crucial.

Recently the enantioselective addition reactions of dialkylzinc and Grignard reagents to a nitrone, 3,4-dihydroisoquinoline *N*-oxide, utilizing bromomagnesium (2*S*,3*R*)-4-dimethylamino-1,2-diphenyl-3-methyl-2-butoxide as a chiral auxiliary was developed to give each enantiomer of the corresponding hydroxylamines, respectively.¹ With these results in hand, we were interested in a catalytic asymmetric addition of organometallics to carbon-nitrogen double bond. In contrast to many successes in the catalytic asymmetric addition of organometallic reagents to carbonyl compounds,² examples of the catalytic asymmetric addition of organometallics to the imine function are still limited.³ Herein, we would like to report our studies on the catalytic asymmetric addition reaction of dialkylzinc to nitrones possessing dihydroisoquinoline skeleton.⁴

The reaction of 2.2 molar amounts of diethylzinc with 3,4-dihydroisoquinoline *N*-oxide **1A**⁵ in the presence of 1.1 molar amounts of bromomagnesium (2*S*,3*R*)-4-dimethylamino-1,2-diphenyl-3-methyl-2-butoxide **2**, prepared from Chiralol^{®6} and Grignard reagent⁷ *in situ*, in THF at 25 °C gave the corresponding hydroxylamine **3a** in 65% ee (Entry 1 in Table 1).⁸ In the reaction using 0.2 molar amounts of magnesium alkoxide **2**, however, optical yield of the hydroxylamine disappointingly lowered to 33% ee (Entry 2). After several attempts, it was found that addition of another magnesium alkoxide generated from the corresponding alcohol and Grignard reagent⁷ *in situ* was effective. That is, addition of 0.2 molar amounts of bromomagnesium butoxide improved the enantioselectivity up to 45% ee (Entry 3). Among alkoxides derived from several alcohols, bromomagnesium triphenylmethoxide **4** was revealed to be the most effective for

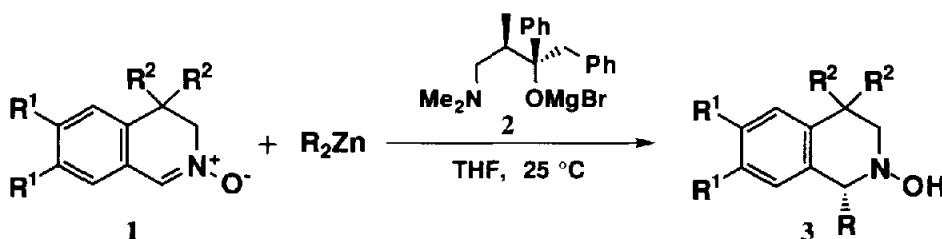
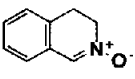
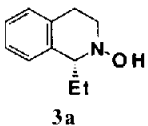
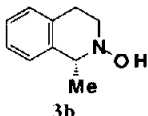
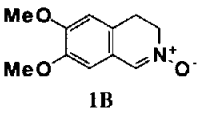
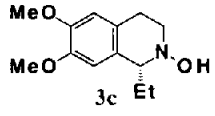
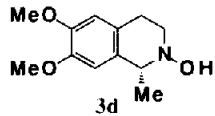
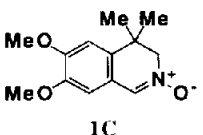
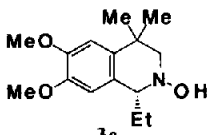


Table 1. The asymmetric addition reaction of dialkylzinc to nitrones **1** using catalytic amount of **2**

Entry	Nitrones 1	Molar amounts of 2	Additive (molar amounts)	R ₂ Zn ^{a)}	Time /h	Products 3	Yield /%	ee ^{b)} /%
1		1.1	---	Et ₂ Zn	13		78	65
2		0.2	---	Et ₂ Zn	16		80	33
3		0.2	<i>n</i> BuOMgBr (0.2)	Et ₂ Zn	17		84	45
4		0.2	<i>i</i> PrOMgBr (0.2)	Et ₂ Zn	17		84	47
5		0.2	<i>t</i> BuOMgBr (0.2)	Et ₂ Zn	17		86	43
6		0.2	Ph ₃ COMgBr (4) (0.2)	Et ₂ Zn	17		82	56
7		0.2	Ph ₃ COMgBr (4) (0.3)	Et ₂ Zn	14		91	62
8		0.2	Ph ₃ COMgBr (4) (0.4)	Et ₂ Zn	24		87	56
9	1A	1.1	---	Me ₂ Zn	17		88	66
10		0.2	---	Me ₂ Zn	17		77	21
11		0.2	Ph ₃ COMgBr (4) (0.3)	Me ₂ Zn	17		93	58
12		1.1	---	Et ₂ Zn	18		87	58
13		0.2	---	Et ₂ Zn	16		60	18
14		0.2	Ph ₃ COMgBr (4) (0.3)	Et ₂ Zn	16		97	70
15	1B	1.1	---	Me ₂ Zn	17		96	50
16		0.2	---	Me ₂ Zn	19		91	57
17		0.2	Ph ₃ COMgBr (4) (0.3)	Me ₂ Zn	16		84	63
18		1.1	---	Et ₂ Zn	15		96	57 ^{c)}
19		0.2	---	Et ₂ Zn	19		80	28 ^{c)}
20		0.2	Ph ₃ COMgBr (4) (0.3)	Et ₂ Zn	18		89	78 ^{c)}

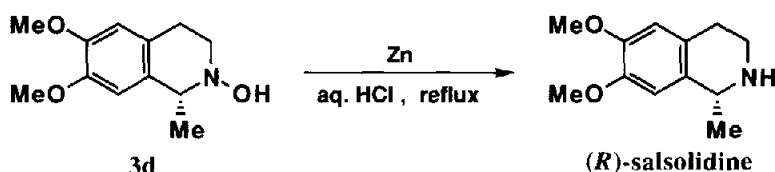
a) 2.2 Molar amounts of dialkylzinc were used. b) Enantiomeric excess was determined by HPLC analysis (Daicel Chiralcel OD-H). c) Enantiomeric excess was determined by ¹H NMR analysis of the corresponding (*R*)- α -methoxy- α -(trifluoromethyl)phenylacetic acid (MTPA) ester derivatives.

the stereoselection and gave the hydroxylamine **3a** in 56% ee (Entries 3-6). Furthermore, the addition of 0.3 molar amounts of **4** improved the stereoselectivity up to 62% ee (Entries 6-8). The reaction of dimethylzinc with the nitrone **1A** was also examined and the similar tendency in stereochemical course was observed (Entries 9-11).

Next, asymmetric addition reaction of dialkylzinc to 6,7-dimethoxy-3,4-dihydroisoquinoline *N*-oxide **1B** was examined. Stoichiometric reaction with diethylzinc afforded the corresponding hydroxylamine **3c** in 58% ee (Entry 12), while the catalytic reaction gave **3c** with rather low stereoselectivity (Entry 13). In

contrast, addition of 0.3 molar amounts of triphenylmethoxide **4** realized higher stereoselectivity than that in the stoichiometric reaction (Entry 14). The reaction of dimethylzinc with **1B** afforded the methylated product **3d** in 63% ee (Entry 17). In the case of the nitron **1C**, the dramatic effect of the additive **4** was also observed (Entries 18-20) and the hydroxylamine **3e** was obtained in 78% ee (Entry 20). These results are summarized in Table 1.

The stereochemistry of the newly formed chiral center in **3d** was determined by conversion to salsolidine;^{4b,9} *i.e.*, reduction⁵ of the hydroxylamine **3d** (50% ee) obtained by the reaction of **1B** with Me₂Zn gave salsolidine ($[\alpha]_D^{25} +31$ (c 1.89, EtOH)) in 92% yield, whose configuration was confirmed to be *R* by the comparison of its specific rotation with that reported for (*R*)-salsolidine ($[\alpha]_D^{22} +62.8$ (c 0.1, EtOH)).^{9e} The stereochemistry of the stereogenic center in **3b** had been already determined to be *R*.¹



Although the precise mechanism of the present reaction is still an open question, the role of bromomagnesium triphenylmethoxide **4** might be explained as follows: Dialkylzinc would react with the nitron coordinated to magnesium alkoxide of ChiralD® (depicted as **5**) from the less hindered *re*-face to afford **6**.¹ In the catalytic system, **2** should be regenerated accompanied with the production of **8**. Without triphenylmethoxide **4**, the rate of regeneration of **2** might be slow and dialkylzinc would react with nitron uncoordinated to **2** without stereoselection. In the presence of **4**, the alkylated adduct **6** would react with alkoxide **4** to give **7** and chiral auxiliary **2** would be smoothly regenerated (Scheme 1). Furthermore, it is also probable that magnesium in triphenylmethoxide **4** might be also coordinated by oxygen of nitron to produce the complex **9**. The *si*-face of nitron **9** would be effectively shielded and alkylation would proceed from *re*-face stereoselectively (Fig. 1 shows the complex **9** corresponding to **1A**) and the adduct immediately dissociates into **2** and **7** followed by reproduction of **9**.

