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## **Asymmetric Addition Reaction of Organozinc Reagents to Nitrones Using a Catalytic Amount of External Chiral Auxiliary**

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Abstract: Catalytic asymmetric addition reactions of dialkylzinc to nitrones are realized; *i.e.,* in the presence of a catalytic amount of bromomagnesium *(2S,3R)-4-dimethylamino-*1,2-diphenyl-3-methyl-2-butoxide, dialkylzincs were reacted with 3,4-dihydroisoquinoline N-oxide derivatives to give the corresponding 1-alkylated hydroxylamines enantioselectively. In order to achieve higher stereoselection, addition of bromomagnesium triphenylmethoxide was crucial.

Recently the enantioselective addition reactions of dialkylzinc and Grignard reagents to a nitrone, 3,4 dihydroisoquinoline N-oxide, utilizing bromomagnesium *(2S,3R)-4-dimethylamino-* 1,2-diphenyl-3-methyl-2 butoxide as a chiral auxiliary was developed to give each enantiomer of the corresponding hydroxylamines, respectively.<sup>1</sup> With these results in hand, we were interested in a catalytic asymmetric addition of organometallics to carbon-nitrogen double bond. In contrast to many successes in the catalytic asymmetric addition of organometallic reagents to carbonyl compounds,<sup>2</sup> examples of the catalytic asymmetric addition of organometallics to the imine function are still limited.<sup>3</sup> Herein, we would like to report our studies on the catalytic asymmetric addition reaction of dialkylzinc to nitrones possessing dihydroisoquinoline skeleton.<sup>4</sup>

The reaction of 2.2 molar amounts of diethylzinc with 3,4-dihydroisoquinoline N-oxide  $1A<sup>5</sup>$  in the presence of 1.1 molar amounts of bromomagnesium *(2S,3R)-4-dimethylamino-l,2-diphenyl-3-methyl-2*  butoxide 2, prepared from Chirald<sup>®6</sup> and Grignard reagent<sup>7</sup> *in situ*, in THF at 25 °C gave the corresponding hydroxylamine 3a in 65% ee (Entry 1 in Table 1).<sup>8</sup> In the reaction using 0.2 molar amounts of magnesium alkoxide 2, however, optical yield of the hydroxylamine disappointingly lowered to 33% ee (Entry 2). After several attempts, it was found that addition of another magnesium alkoxide generated from the corresponding alcohol and Grignard reagent<sup>7</sup> in situ was effective. That is, addition of 0.2 molar amounts of bromomagnesium butoxide improved the enantioselectivity up to 45% ee (Entry 3). Among alkoxides derived from several alcohols, bromomagnesium triphenylmethoxide 4 was revealed to be the most effective for



Entry	Nitrones	Molar	Additive		R <sub>2</sub> Zn <sup>a</sup> ) Time		Products	Yield	ee <sub>b</sub>
	$\mathbf{1}$	amounts of 2	(molar amounts)			$\Lambda$	$\overline{\mathbf{3}}$	$P'$ o	1%
$\mathbf{I}$		1.1			Et <sub>2</sub> Zn	13		78	65
$\overline{2}$		0.2			$E_1/2n$	16		80	33
3		0.2	nBuOMgBr	(0.2)	Et <sub>2</sub> Zn	17		84	45
$\overline{4}$		0.2	<i>i</i> PrOMgBr	(0.2)	Et <sub>2</sub> Zn	17	OН	84	47
5	1A	0.2	tBuOMgBr	(0.2)	Et2Zn	17	Ē١	86	43
6		0.2	Ph3COMgBr $(4)(0.2)$		Et <sub>2</sub> Zn	17	3a	82	56
7		0.2	Ph3COMgBr (4)(0.3) Et2Zn			14		91	62
8		0.2	Ph3COMgBr $(4)(0.4)$ Et2Zn			24		87	56
9		1.1			Me <sub>2</sub> Zn	-17		88	66
10	1A	0.2			Me <sub>2</sub> Zn	17		77	21
11		0.2	Ph3COMgBr $(4)(0.3)$ Me <sub>2</sub> Zn			-17	Мe 3 <sub>b</sub>	93	58
$12 \text{ }$	MeQ.	1.1			Et2Zn	18	MeO.	87	58
13		0.2			Et2Zn	16		60	18
14	MeO <sup>®</sup> 1 <b>B</b>	0.2	Ph3COMgBr (4)(0.3) Et2Zn			16	MeO $3c$ Et	он 97	70
15		1.1			Me <sub>2</sub> Zn	17	MeO.	96	50
16	1B	0.2			Me <sub>2</sub> Zn	19		91	57
17		0.2	Ph3COMgBr $(4)(0.3)$ Me <sub>2</sub> Zn			$-16$	MeO Me 3d	он 84	63
	Me Me	1.1			Et <sub>2</sub> Zn	15	Me Me	96	57c)
18	MeO				Et <sub>2</sub> Zn	19	MeO	80	28c
19	MeO	0.2 0.2			Et <sub>2</sub> Zn	18	MeO	он 89	78c)
20	1 <sup>C</sup>		Ph3COMgBr $(4)(0.3)$				Ēt 3 <sub>e</sub>		

Table 1. The asymmetric addition reaction of dialkylzinc to nitrones 1 using catalytic amount of 2

a) 2.2 Molar amounts of dialkylzinc were used. b) Enantiomeric excess was determined by HPLC analysis (Daicel Chiralcel OD-H). c) Enantiomeric excess was determined by <sup>1</sup>H NMR analysis of the corresponding  $(R)$ - $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetic acid (MTPA) ester derivatives.

the stereoselection and gave the hydroxylamine 3a in 56% ee (Entries 3-6). Furthermore, the addition of 0.3 molar amounts of 4 improved the stereoselectivity up to 62% ee (Entries 6-8). The reaction of dimethylzinc with the nitrone 1A was also examined and the similar tendency in stereochemical course was observed (Entries 9-11).

Next, asymmetric addition reaction of dialkylzinc to 6,7-dimethoxy-3,4-dihydroisoquinoline N-oxide IB was examined. Stoichiometric reaction with diethylzinc afforded the corresponding hydroxylamine 3c in 58% ee (Entry 12), while the catalytic reaction gave 3c with rather low stereoselectivity (Entry 13). In contrast, addition of 0.3 molar amounts of triphenylmethoxide 4 realized higher stereoselectivity than that in the stoichiometric reaction (Entry 14). The reaction of dimethylzinc with 1B afforded the methylated product 3d in  $63\%$  ee (Entry 17). In the case of the nitrone 1C, the dramatic effect of the additive 4 was also observed (Entries 18-20) and the hydroxylamine 3e was obtained in 78% ee (Entry 20). These results are summarized in Table 1.

The stereochemistry of the newly formed chiral center in 3d was determined by conversion to salsolidine;<sup>4b, 9</sup> *i.e.*, reduction<sup>5</sup> of the hydroxylamine 3d (50% ee) obtained by the reaction of 1B with Me<sub>2</sub>Zn gave salsolidine ( $\alpha$ ]<sub>D</sub><sup>25</sup> +31 (c 1.89, EtOH)) in 92% yield, whose configuration was confirmed to be R by the comparison of its specific rotation with that reported for (R)-salsolidine ( $\left[\alpha\right]_D^2$  +62.8 (c 0.1, EtOH)). <sup>9e</sup> The stereochemistry of the stereogenic center in 3b had been already determined to be  $R<sup>1</sup>$ 



Although the precise mechanism of the present reaction is still an open question, the role of bromomagnesium triphenylmethoxide 4 might be explained as follows: Dialkylzinc would react with the nitrone coordinated to magnesium alkoxide of Chirald<sup>®</sup> (depicted as 5) from the less hindered re-face to afford  $6<sup>1</sup>$  In the catalytic system, 2 should be regenerated accompanied with the production of 8. Without triphenylmethoxide 4, the rate of regeneration of 2 might be slow and dialkylzinc would react with nitrone uncoordinated to 2 without stereoselection. In the presence of 4, the alkylated adduct  $6$  would react with alkoxide 4 to give 7 and chiral auxiliary 2 would be smoothly regenerated (Scheme 1). Furthermore, it is also probable that magnesium in triphenylmethoxide 4 might be also coordinated by oxygen of nitrone to produce the complex 9. The si-face of nitrone 9 would be effectively shielded and alkylation would proceed from *re*face stereoselectively (Fig. 1 shows the complex 9 corresponding to 1A) and the adduct immediately dissociates into 2 and 7 followed by reproduction of 9.





In summary, we have demonstrated the catalytic asymmetric addition reaction of dialkylzinc to carbonnitrogen double bond and the present reaction provided a facile method for the synthesis optically active 1 alkyltetrahydroisoquinoline alkaloids such as (R)-salsolidine.

References and Notes

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